

Name: _____ Block: _____ Date: _____

Homework – Significant Figures and Scientific Notation

Determine the number of significant figures for each of the following.

- | | | | | | | | |
|-----------|----------|-----------|----------|------------|----------|--------------|----------|
| 1. 5.432 | _____ | 6. 40.319 | <u>5</u> | 11. 146 | _____ | 16. 3.285 | <u>4</u> |
| 2. 0.189 | <u>3</u> | 7. 429.3 | _____ | 12. 2873.0 | <u>5</u> | 17. 99.9 | _____ |
| 3. 0.0023 | _____ | 8. 144 | <u>3</u> | 13. 2500 | _____ | 18. 2500.0 | <u>5</u> |
| 4. 1.04 | <u>3</u> | 9. 35.08 | _____ | 14. 8365.6 | <u>5</u> | 19. 48.57193 | _____ |
| 5. 7.500 | _____ | 10. 7,500 | <u>2</u> | 15. 0.920 | _____ | 20. 0.002300 | <u>4</u> |

Using significant figures, calculate the following addition and subtraction problems.

- $12\text{ cm} + 0.031\text{ cm} + 7.969\text{ cm} =$ _____
- $0.085\text{ cm} + 0.062\text{ cm} + 0.14\text{ cm} =$ 0.29 cm
- $3.419\text{ g} + 3.912\text{ g} + 7.0518\text{ g} + 0.00013\text{ g} =$ _____
- $8.7\text{ g} + 15.43\text{ g} + 19\text{ g} =$ 43g
- $143.0\text{ ml} + 289.25\text{ ml} + 107.85\text{ ml} =$ _____
- $41.025\text{ cm} - 23.28\text{ cm} =$ 17.75 cm
- $289\text{ g} - 43.7\text{ g} =$ _____
- $145.63\text{ ml} - 28.9\text{ ml} =$ 116.7 ml
- $62.47\text{ g} - 39.9\text{ g} =$ _____
- $40.08\text{ ml} - 29.0941\text{ ml} =$ 10.99 ml

+/- use the
number line
Hundreds
Tens
ones
Tenths
Hundredths

Positional

Using significant figures, calculate the following multiplication and division problems.

- $2.89\text{ cm} \times 4.01\text{ cm} =$ _____
- $17.3\text{ cm} \times 6.2\text{ cm} =$ 110 cm²
- $3.08\text{ m} \times 1.2\text{ m} =$ _____
- $5.00\text{ mm} \times 7.3216\text{ mm} =$ 36.6 mm²
- $20.8\text{ dm} \times 123.1\text{ dm} =$ _____
- $8.071\text{ cm}^2 \div 4.216\text{ cm} =$ 1.914 cm
- $24,789.4\text{ km}^2 \div 43.5\text{ km} =$ _____
- $109.3758\text{ m}^2 \div 5.813\text{ m} =$ 18.82 m
- $6.058\text{ mm}^2 \div 0.85\text{ mm} =$ _____
- $4.23\text{ m}^2 \div 18,491\text{ m} =$ 0.229 m

Convert the following standard notations to scientific notation.

- | | | | |
|----------------|--|---------------|---------------------------------------|
| 1. 28,000,000 | _____ | 6. 62,500 | <u>6.25×10^4</u> |
| 2. 305,000 | <u>3.05×10^5</u> | 7. 0.002403 | _____ |
| 3. 0.000024863 | _____ | 8. 8,809,000 | <u>8.809×10^6</u> |
| 4. 345.23 | <u>3.4523×10^2</u> | 9. 0.251 | _____ |
| → 5. 0.00025 | <u>2.5×10^{-4}</u> | 10. 3,010,000 | <u>3.01×10^6</u> |

9/25/15 Communication Through Numbers

Accuracy: ability to get close to the expected result.

Precision: ability to repeat results - consistency

$$\% \text{ error} = \left| \frac{(\text{measured value} - \text{accepted value})}{\text{accepted value}} \right| \times 100 =$$

↑ (measures accuracy)

measuring precision = results w/ the smallest range.

highest value - lowest value = range

Practice

$$\textcircled{4} \% \text{ error} = \left| \frac{(m - A)}{A} \right| \times 100 \quad A = 47.00\text{g}$$

$$A \% \text{ error} = \left| \frac{(47.01\text{g} - 47.00\text{g})}{47.00\text{g}} \right| \times 100 = 0.02128 \% \text{ error} \leftarrow \text{most accurate}$$

$$B \% \text{ error} = \left| \frac{(46.88\text{g} - 47.00\text{g})}{47.00\text{g}} \right| \times 100 = 0.2553 \% \text{ error}$$