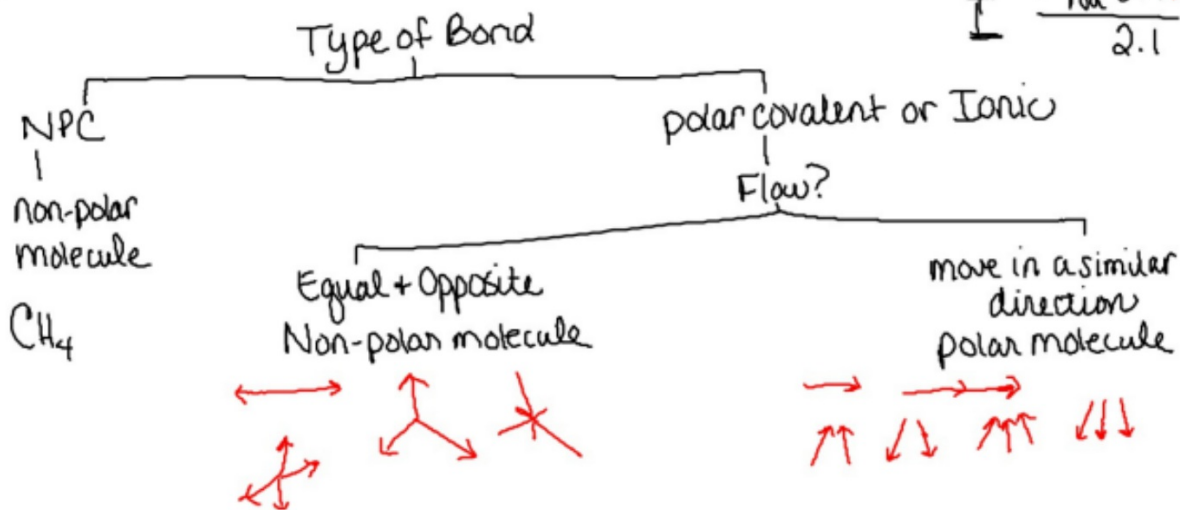
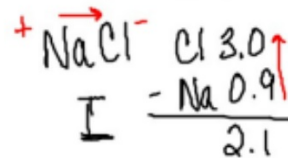
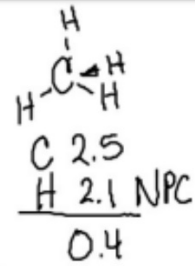
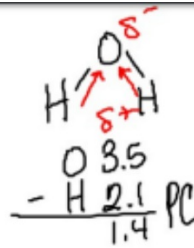
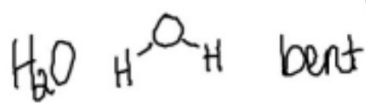
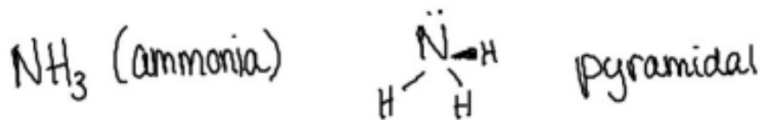


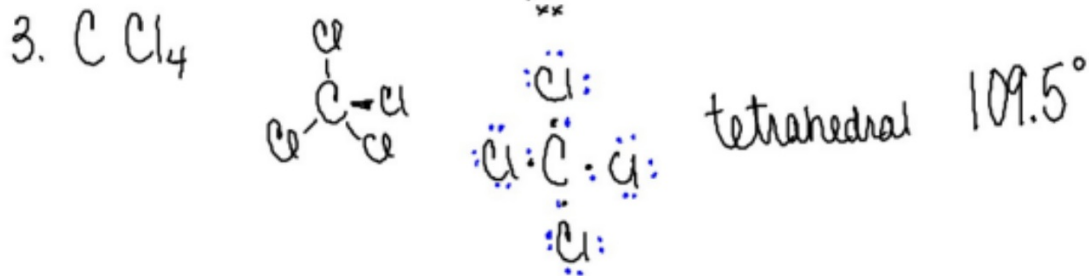
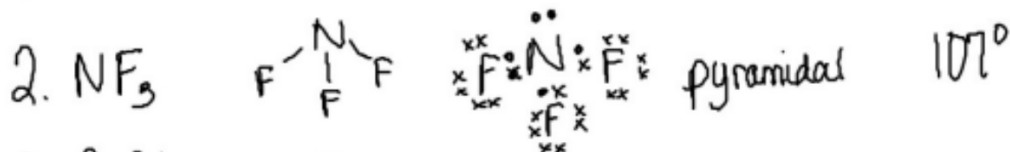
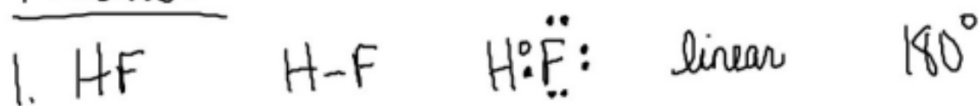
## Determining Molecular Polarity

1. Calculate what types of bonds are present
2. Place arrows on the bonds if pc or ionic - the arrows must face the ↑ EN.
3. Evaluate "Flow" of arrows





Practice:



Memory Work

acetate	$C_2H_3O_2^{1-}$
chlorate	$ClO_3^{1-}$
nitrate	$NO_3^{1-}$
carbonate	$CO_3^{2-}$
sulfate	$SO_4^{2-}$
phosphate	$PO_4^{3-}$
hydroxide	$OH^{1-}$
cyanide	$CN^{1-}$
peroxide	$O_2^{2-}$
ammonium	$NH_4^{1+}$