

Homework:

2.19 Pairs of constitutional isomers b,c,e + f

2.21 a) ethyl b) isopropyl c) isobutyl d) tert-butyl (t-butyl)

2.23 a) 2-methylpentane

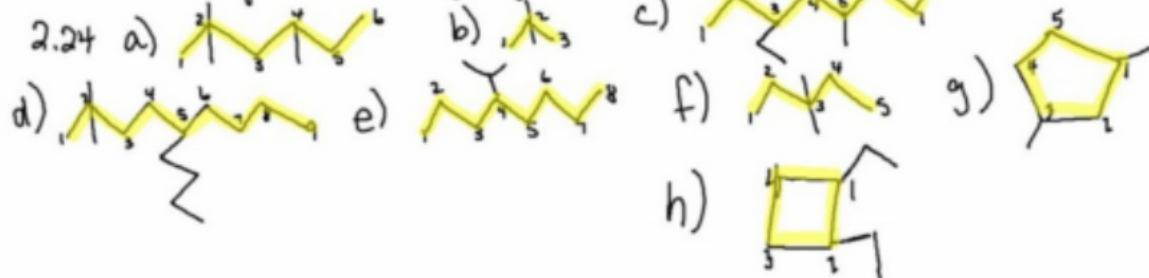
b) 2,5-dimethyl hexane

c) 3-ethyl octane

d) 1-isopropyl-2-methyl cyclohexane

e) isobutyl cyclopentane

f) 1-ethyl-2,4-dimethyl cyclohexane



2.25 A conformation is any 3-D arrangement of the atoms in a molecule that result from the rotation around a single bond.

2.26 The carbon-carbon bonds actually zig-zag because of the 109.5° bond angle.

2.28 when a hydrocarbon is in a ring formation the bonds are locked into position allowing for cis-trans formation (geometric isomer)

ΔT Propanol	ΔT Isopropanol	ΔT Acetone	ΔT Acetic Acid
3.3°C	3.8°C	17.0°C	2.9°C
6.4°C	7.5°C	17.4°C	3.3°C
4.5°C	6.5°C	13.4°C	3.6°C
6.1°C	3.9°C	12.7°C	2.9°C
4.3°C	5.5°C	14.3°C	3.6°C
3.5°	5.5°C	11.8°C	3.8°C
Average 4.7°C	5.5°C	14.4 °C	3.4°C

ΔT Propanol	ΔT Isopropanol	ΔT Acetone	ΔT Acetic Acid
0.7°C	1.6°C	6.7°C	0.7°C
3.3°C	6.2°C	9.6°C	3.8°C
3.7°C	5.4°C	12.5°C	4.0°C
4.2°C	5.8°C	12.3°C	3.1°C
4.0°C	6.0°C	17.2°C	5.2°C
5.3°C	7.2°C	13.2°C	4.1°C
5.1°C	7.5°C	9.7°C	4.3°C
5.8°C	7.5°C	17.5°C	3.9°C
3.2°C	3.8°C	11.4°C	4.7°C
4.5°C	6.4°C	14.5°C	4.1°C

Avg:
prop 4.0°C
isoprop 5.7°C
acetone 12.5°C
acetic acid 3.8°C

