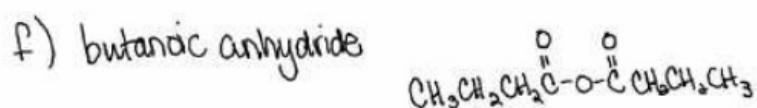
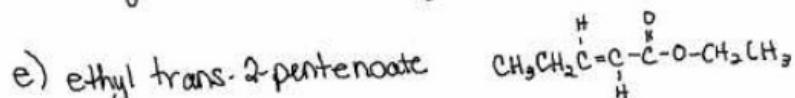
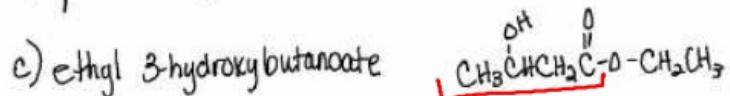
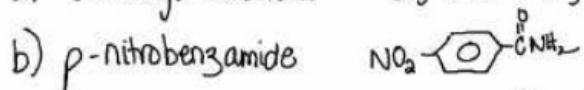
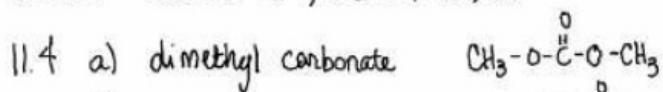
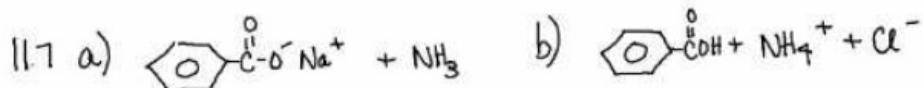


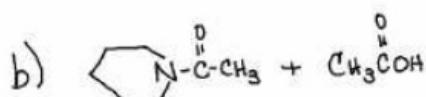
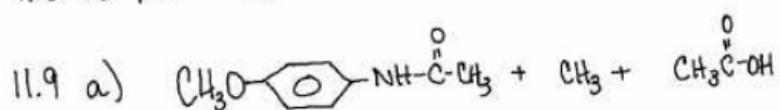
1/8/13 Homework: 11.4, 11.5, 11.7, 11.8, 11.9

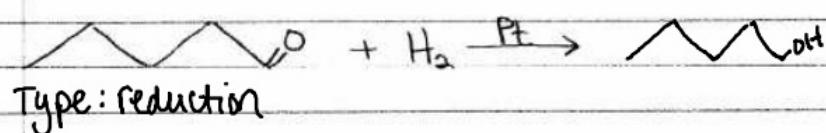
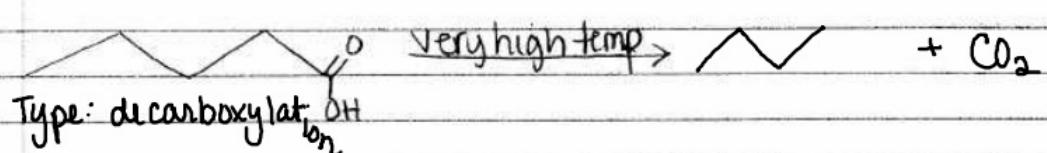
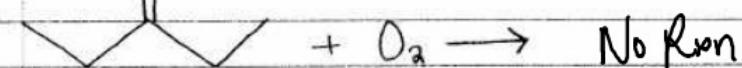
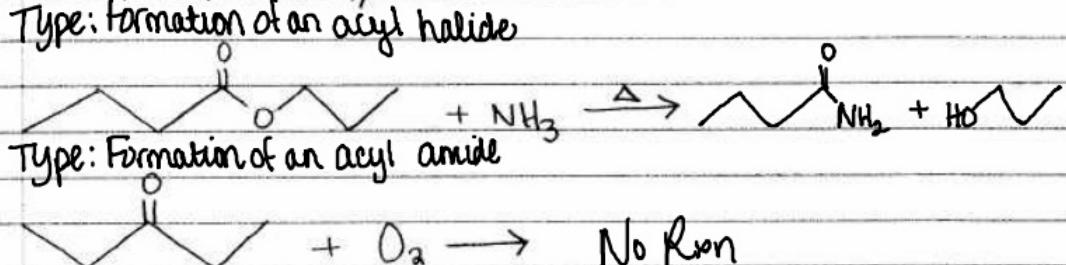
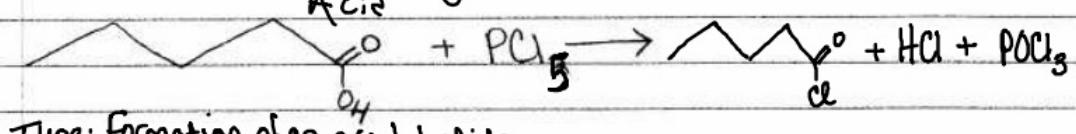
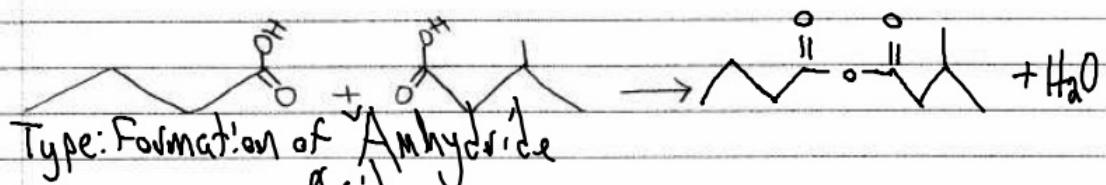
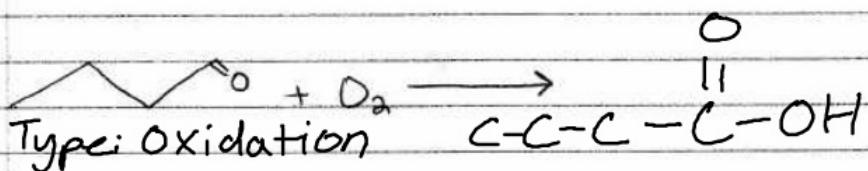
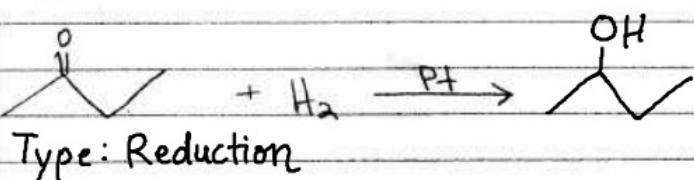
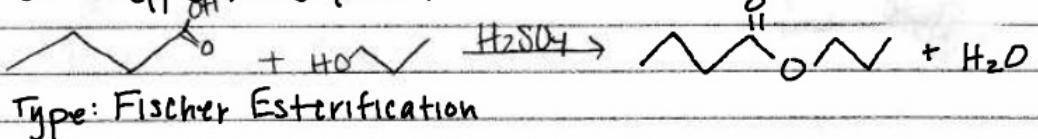
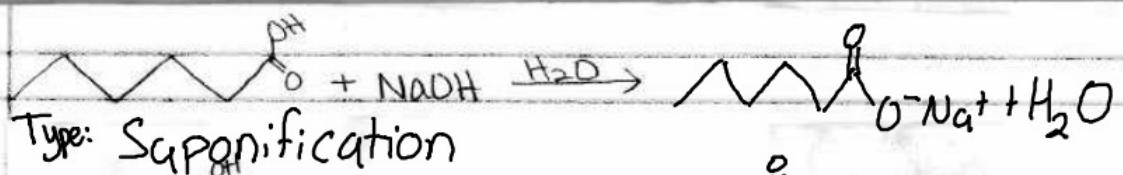


- 11.5 a) benzoic anhydride b) methyl decanoate c) N-Methyl hexanamide  
d) <sup>(+)</sup> *p*-amino benzamide e) cyclopentyl ethanoate f) ethyl 3-hydroxybutanoate



- 11.8 a) Yes b) No c) No







Type: basicity of an amine



Type: neutralization of an amine

1 A	10 A	19 D	28 B	39 C	#5-excluding cyclic ketones
2 C	11 A	20 B	29 B	40 C	
3 B	12 C	<u>21 A</u>	30 B	41 B	
4 C	13 C	<u>22 B</u>	<u>31 B</u>	<u>42 B</u>	
5 C	14 B	23 B	32 D	43 A	
6 C	15 A	24 B	33 C	44 B	
7 A	16 B	25 B	34 D	45 C	
8 B	17 B	26 A	35 B	46 D	
9 C/D	18 D	27 A	36 C	47 D	
			37 B	48 A	
			38 B	49 B	
				50 A	

For Unit 5/6 Test (Ch 8,9,10,11)

50 m.c. questions

Name / Illustrate :

Aldehydes

ketones

Carboxylic Acids

Amines (also common names)

amides

acid anhydrides

(esters)

Reactions : (matching)

oxidation

reduction

formation of :

esterification

acyl halide

acyl amide

Neutralization

acid anhydride

decarboxylation

hydrolysis

basicity of amine